

Phenylselenyl Halide Induced Formation of Cyclic Nitrones from Alkenyl Oximes

Ronald Grigg,*^a Maria Hadjisoteriou,^a Peter Kennewell^b and Jasothara Markandu^a

^a School of Chemistry, Leeds University, Leeds LS2 9JT, UK

^b Roussel Scientific Institute, Kingfisher Drive, Swindon SN3 5BZ, UK

Oximes possessing alk- γ - or - δ -enyl substituents are cyclised by phenylselenyl bromide, or by phenylselenyl chloride and an appropriate silver salt, to the corresponding cyclic nitrones; the seleno nitrones undergo facially specific cycloadditions with *N*-methylmaleimide, bis(alk- γ , δ -enyl) and bis(alk- δ , δ -enyl) ketones undergo regiospecific cyclisation and stereospecific cycloaddition to furnish spirocyclic products.

Inter- and intra-molecular cycloaddition reactions of nitrones have attracted much attention because they provide a potentially flexible entry into the complex molecular frameworks of natural products.¹ Utilising oximes as nitronone precursors in tandem nitronone generation-cycloaddition protocols substantially enhances this flexibility. We have recently introduced a range of such protocols² most of which have four distinct synthetic variants. Further novel developments in this area are now reported.

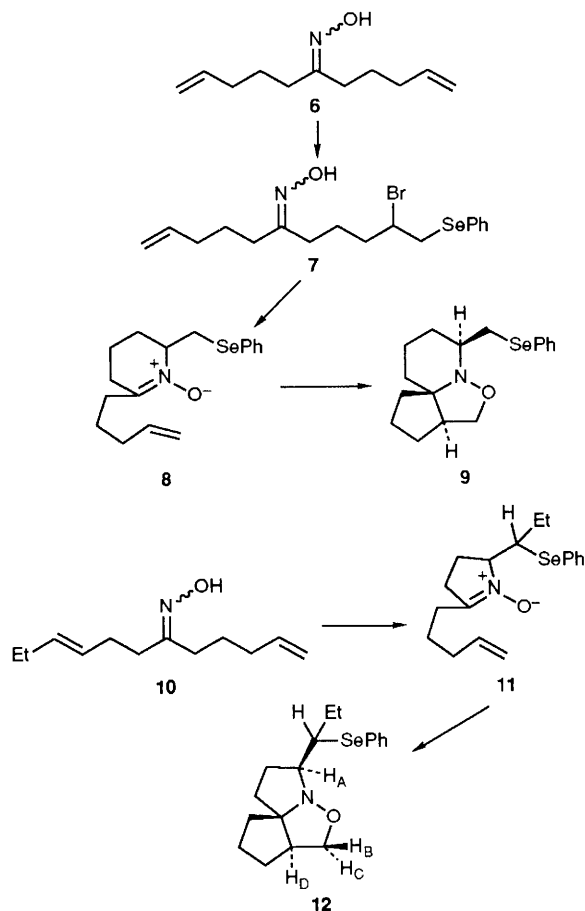
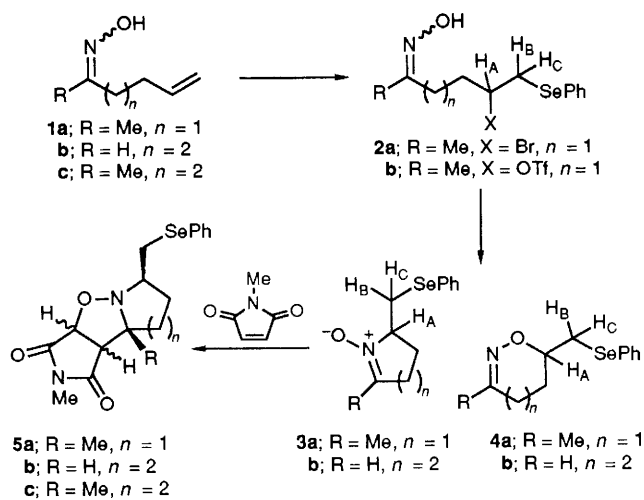
We have been exploring the generality of electrophile induced oxime-olefin (alkyne) reactions as a source of novel tandem nitronone formation-cycloaddition protocols and in our preceding communication² we reported reactions in which mercuric acetate functioned as the electrophile. This paper is concerned with the application of phenylselenyl halides³ to such processes.

In a preliminary evaluation of phenylselenyl halides it became clear that phenylselenyl bromide, or phenylselenyl chloride in combination with silver triflate (trifluoromethanesulfonate), usually gave rise to cleaner reactions and improved yields compared with phenylselenyl chloride alone or phenylselenyl chloride in combination with silver tetrafluoroborate.

Oxime **1a** which comprised a 2 : 1 mixture of *syn* (Me/OH)- and *anti*-isomers reacted (CH₂Cl₂, 25 °C, 0.5 h) with phenylselenyl bromide (1 mol) to give a quantitative yield of the adduct **2a**.[†] Treatment of **2a** with anhydrous potassium carbonate (1.1 mol) (CH₂Cl₂, 25 °C, 16 h) afforded the nitronone **3a** together with a trace of oxazine **4a**.[‡] Heating the nitronone **3a** in acetonitrile (80 °C, 9 h) with *N*-methylmaleimide (1 mol) afforded cycloadduct **5a** (70% overall from **1a**) as a 3 : 2 mixture of *endo*- and *exo*-isomers. It is apparent from this result that the ratio of nitronone **3a** to oxazine **4a** does not reflect the *syn* : *anti* ratio of the oxime. The nitronone : oxazine ratio is sensitive to the reagents/conditions used as shown by reaction of **1a** with PhSeCl-AgOTf (CH₂Cl₂, 0 °C) to give **2b**[†] followed

[†] The ¹H NMR spectrum (CDCl₃) of **2a** exhibited a broad singlet at δ 5.1 (H_A) and two double doublets at δ 3.6 and 3.34 (H_B, H_C); **2b** δ 5.0 (br s, 1H, H_A) and 3.55 and 3.45 (2 \times dd, 2 \times 1H, H_B, H_C).

[‡] ¹H NMR data (CDCl₃): **3a** δ 4.3 (br s, 1H, H_A), 3.62 and 3.22 (2 \times dd, 2 \times 1H, H_B, H_C) and 2.0 (s, 3H, Me); **4a** δ 3.8 (m, 1H, H_A), 3.25 and 2.9 (2 \times dd, 2 \times 1H, H_B, H_C) and 1.9 (s, 2H, Me).



by treatment with potassium carbonate (CH_2Cl_2 , 25 °C), which affords a 2.5:1 mixture of **3a** and **4a** in 76% combined yield. The diastereofacially specific cycloaddition of this and

all the nitrones in this paper accords with our observations on related cases.^{2,4}

Oxime **1b** (1:1 mixture of *syn*- and *anti*-isomers) reacted under analogous conditions to give the nitron **3b** and subsequently the cycloadducts **5b** (61% overall from **1b**). In this case none of the corresponding oxazine **4b** was observed and only the *endo*-cycloadduct **5b** was obtained. The oxime **1c** reacted under the same conditions to give cycloadduct **5c** as a 3:2 mixture of *endo*- and *exo*-cycloadducts but again no oxazine was detected. The conversion of oximes **1a-c** into cycloadducts **5a-c** constitute examples of Class 3 processes according to our classification.²

Class 4 processes² have been explored with both symmetrical and unsymmetrical substrates. Thus, **6** reacts (MeCN , 25 °C, 0.5 h) with phenylselenenyl bromide to afford **7**, which on treatment with anhydrous potassium carbonate (1.1 mol) (MeCN , 25 °C, 16 h) affords nitron **8**. The nitron on heating (MeCN , 80 °C, 5 h) affords a single spirocyclic cycloadduct **9** (65% overall from **6**).

A similar sequence of reactions was performed on the unsymmetrical dialk- γ,δ -enyl oximes **10**. Regiospecific cyclisation (MeCN , 25 °C) to nitron **11** occurred on treatment of **10** sequentially with phenylselenenyl bromide and potassium carbonate. Heating of **11** in acetonitrile (80 °C, 4 h) afforded a single cycloadduct **12** (61%) via the anticipated diastereofacially specific cycloaddition.

The stereochemistry of the cycloadducts reported herein was established by PMR decoupling experiments and NOE studies. A typical example is provided by **12** in which irradiation of H_C caused enhancements of the signals for H_B , H_A and H_D .

Further work on these and related electrophile induced conversion of oximes to nitrones is underway.

We thank the SERC, Roussel and Leeds University for support.

Received, 3rd August 1992; Com. 2/04157J

References

- 1 J. J. Tufariello, in *1,3-Dipolar Cycloaddition Chemistry*, ed. A. Padwa, Wiley, New York, 1984, vol. 12, p. 83; K. B. G. Torrsell, *Nitrile Oxides, Nitrones and Nitronates in Organic Synthesis*, VCH, Weinheim, 1988.
- 2 These are summarised in, R. Grigg, M. Hadjisoteriou, P. Kennewell, J. Markandu and M. Thornton-Pett, *J. Chem. Soc., Chem. Commun.*, 1992, 1388.
- 3 For leading references see: C. Paulmier, in *Selenium Reagents and Intermediates in Organic Synthesis*, Pergamon Press, Oxford, 1986, ch. 8.
- 4 R. Grigg, J. Markandu, T. Perrior, S. Surendrakumar and W. J. Warnock, *Tetrahedron*, 1992, **48**, 6929; R. Grigg, J. Markandu, S. Surendrakumar, M. Thornton-Pett and W. J. Warnock, *Tetrahedron*, in the press.